

Measuring Density

Organise the method used to measure density

- Calculate and record the volumes (length, width, height).
- Record your results in a table.
- For each object measure the: length, width, height.
- Record the results.
- Calculate and record the densities (mass ÷ volume).
- Include columns for volume, mass, density and substance.
- Measure the mass of each object using the digital balance.

Describe how adding sugar to water affects the density.

Mass of sugar dissolved in 0.1Kg of water (kg)	Density (kg/m ³)
0.005kg	1000 kg/m ³
0.01g	1005 kg/m ³
0.015g	1007 kg/m ³
0.02g	1009 kg/m ³
0.025g	1012 kg/m ³

Why do we not take into account the volume of the sugar?



How should the method be modified for measuring the density of a liquid. Write extra instructions below:

Density

Formula

$$\text{Density} = \frac{\text{Mass}}{\text{Volume}}$$



Rearrange this formula to find:

Mass = _____

Volume = _____

What are the units?

Mass is measured in _____

Volume is measured in _____

Density is measured in _____

Plan

Without turning over (!) write a step by step plan for measuring the density of a solid.

Convert the following units

1. 500g = Kg
2. 25g = Kg
3. 770g = Kg
4. 58g = Kg
5. 10,000cm³ = m³
6. 100cm³ = m³
7. 250,000cm³ = m³
8. 100,000cm³ = m³



Help?

$$1000\text{g} = 1\text{kg}$$

$$\text{g} \rightarrow \text{kg} \div 1000$$

$$\text{kg} \rightarrow \text{g} \times 1000$$

$$1\text{ m}^3 = 1,000,000\text{cm}^3$$

$$\text{m}^3 \rightarrow \text{cm}^3 \times 1,000,000$$

$$\text{cm}^3 \rightarrow \text{m}^3 \div 1,000,000$$

Confident?

a. 2g = Kg

b. 34.5g =kg

c. 0.5 g =kg

d. 10 cm³ = m³

e. 198 cm³ = m³

f. 2.5 cm³ = m³

Calculations:

1. A solid block has dimensions of 100 cm x 100cm x 100 cm and a mass of 500g. Calculate its density.
2. A solid block has dimensions of 12 cm x 8 cm x 5 cm and a mass of 500g. Calculate its density.
3. A solid block has dimensions of 6 cm x 8 cm x 4.5 cm and a mass of 273g. Calculate its density.